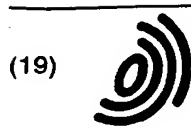


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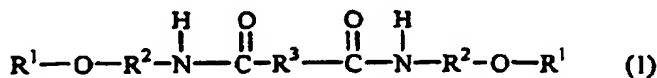
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(54) **EXTERNAL PREPARATION COMPOSITIONS**

(57) The present invention relates to a dermatologic preparation containing a diamide derivative represented by the following formula (1):



(wherein, R¹ represents a linear or branched hydrocarbon group having 1 to 22 carbon atoms which may be substituted by one or more hydroxy and/or alkoxy groups, R² represents a linear or branched divalent hydrocarbon group having 1 to 12 carbon atoms, and R³ represents a linear or branched divalent hydrocarbon group having 1 to 42 carbon atoms). This diamide derivative (1) is capable of fundamentally improving the water retention capacity and barrier functions of the horny layer, is excellent in miscibility and mixing stability and can be prepared efficiently at a low cost.

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Description

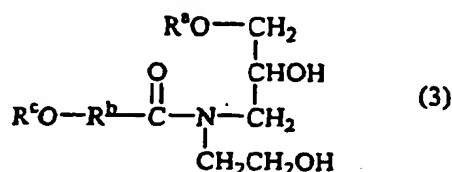
Technical Field

[0001] The present invention relates to dermatologic preparations capable of exerting excellent effects of maintaining normal barrier functions of the horny layer, restoring and reinforcing damaged barrier functions, heightening water retention of the horny layer and remedying skin chapping; and novel diamide derivatives having such effects.

Background Art

[0002] When the water retention capacity or barrier functions of the horny layer are weakened by some internal or external reasons, the skin suffers from troubles such as chapping and acceleration of aging. It is therefore highly important to maintain and reinforce the water retention capacity and barrier functions of the horny layer for our healthy daily life.

[0003] The present applicant has formerly proposed dermatologic preparations comprising an amide derivative represented by the following formula (3):



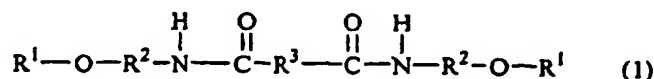
(wherein, R^a represents a linear or branched, saturated or unsaturated hydrocarbon group having 10 to 40 carbon atoms, R^b represents a linear or branched divalent hydrocarbon group having 3 to 39 carbon atoms, R^c represents a hydrogen atom, a linear or branched, saturated or unsaturated hydrocarbon group having 10 to 40 carbon atoms or an acyl group) as a dermatologic preparation capable of essentially improving (maintaining, reinforcing) the barrier functions of the horny layer (Japanese Patent Application Laid-Open (*Kokai*) No. 4-128256).

[0004] Although these amide derivatives exert the above-described excellent effects, they still involve some problems in miscibility or mixing stability because they do not always have sufficient solubility in bases and solution stability. Moreover, preparation of such amide derivatives requires multi-stage reaction, inevitably causing an increase in their production cost.

[0005] An object of the present invention is therefore to provide a compound capable of essentially improving the water retention capacity and barrier functions of the horny layer, having improved miscibility or mixing stability, and being available efficiently at a low cost; and a dermatologic preparation which contains the above-described compound and, by maintaining and reinforcing the water retention capacity and barrier functions of the horny layer, exerts effects of preventing or remedying skin troubles such as chapping, protecting the hair with its penetrated component, improving touch feel of the hair and preventing or remedying chapping of the scalp.

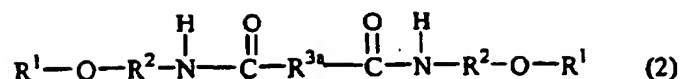
Disclosure of the Invention

[0006] The present invention provides a dermatologic preparation, humectant or skin-barrier-function reinforcing agent, which comprises a diamide derivative represented by the following formula (1):



(wherein R^1 represents, a linear or branched hydrocarbon group having 1 to 22 carbon atoms which may be substituted by one or more hydroxy and/or alkoxy groups, R^2 represents a linear or branched divalent hydrocarbon group having 1 to 12 carbon atoms, and R^3 represents a linear or branched divalent hydrocarbon group having 1 to 42 carbon atoms).

[0007] The present invention also provides a diamide derivative represented by the following formula (2):



(wherein, R^1 represents a linear or branched hydrocarbon group having 1 to 22 carbon atoms which may be substituted by one or more hydroxy and/or alkoxy groups, R^2 represents a linear or branched divalent hydrocarbon group having 1 to 12 carbon atoms, and R^{3a} represents an alkylene group or an alkenylene group having 1 to 4 double bonds, which alkylene or alkenylene group may be linear or branched and has 11 to 42 carbon atoms).

Best Mode for Carrying out the Invention

[0008] In the diamide derivatives (1) and (2), preferred as R^1 are linear or branched C_{1-22} alkyl groups which may have 1 to 3 substituents selected from a hydroxy group and C_{1-6} alkoxy groups. Of these, C_{1-18} alkyl groups, C_{1-18} mono- or di-hydroxyalkyl groups, (C_{1-6} alkoxy-substituted)-(C_{1-18} alkyl) groups, and (hydroxy- and C_{1-6} alkoxy-substituted)-(C_{1-18} alkyl) groups are preferred, of which C_{1-18} alkyl groups, C_{2-12} mono- or di-hydroxyalkyl groups, (C_{1-6} alkoxy-substituted)-(C_{2-12} alkyl) groups, and (hydroxy- and C_{1-6} alkoxy-substituted)-(C_{2-12} alkyl) groups are more preferred. Specific examples include methyl, ethyl, propyl, butyl, hexyl, dodecyl, 2-methylpropyl, 2-ethylhexyl, methyl-branched isostearyl, 2-hydroxyethyl, 9-hydroxynonyl, 2,3-dihydroxypropyl, 2-methoxyethyl, 2-hydroxy-3-methoxypropyl and 9-methoxynonyl groups. Of these, 2-hydroxyethyl, methyl, dodecyl and 2-methoxyethyl groups are more preferred.

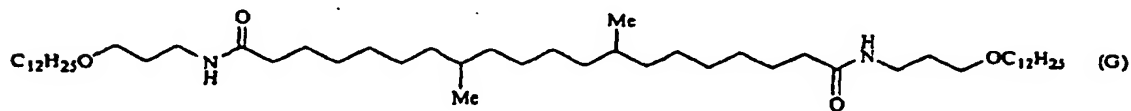
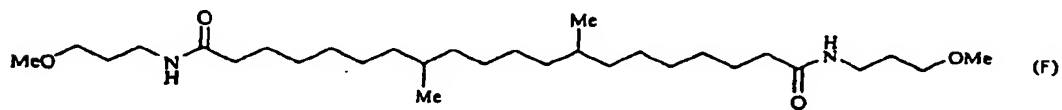
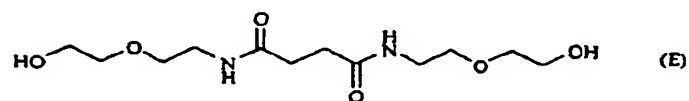
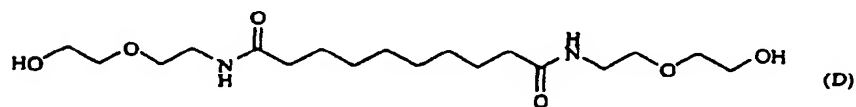
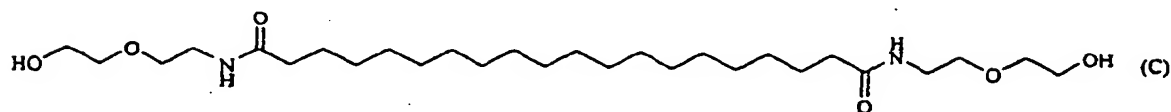
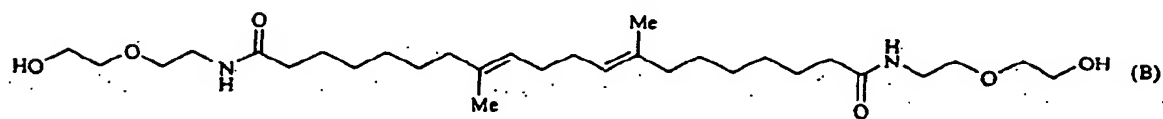
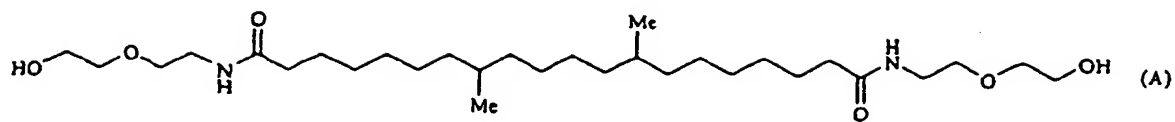
[0009] As R^2 , linear or branched C_{1-12} alkylene groups are preferred, with linear or branched C_{2-6} alkylene groups being more preferred. Specific examples include ethylene, trimethylene, tetramethylene, pentamethylene, hexamethylene, methylmethylene (ethylidene), 1-methylethylene, 2-methylethylene, 1-methyltrimethylene, 2-methyltrimethylene, 1,1-dimethylethylene and 2-ethyltrimethylene groups, of which ethylene and trimethylene groups are more preferred.

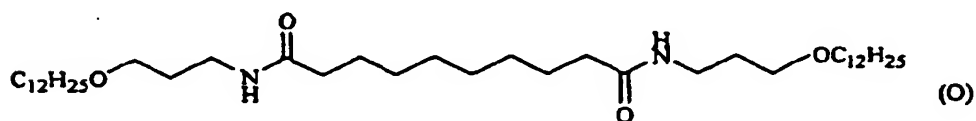
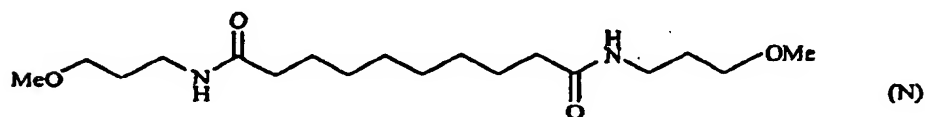
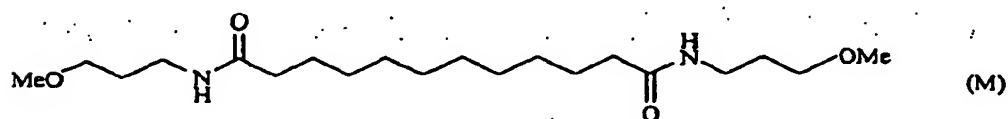
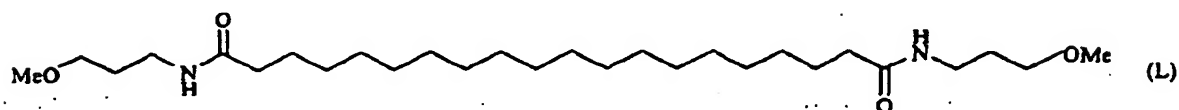
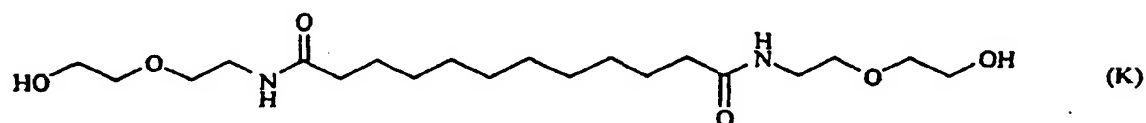
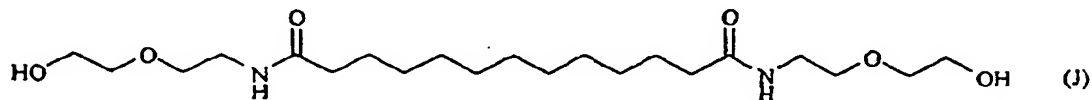
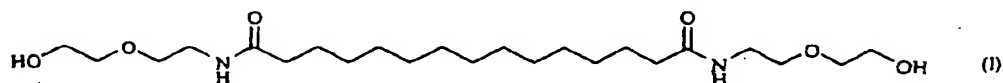
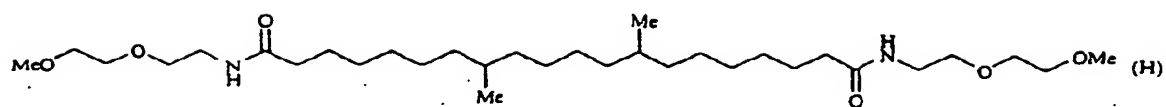
[0010] In the formula (1), linear or branched divalent hydrocarbon groups having 2 to 34 carbon atoms are preferred as R^3 , with alkylene groups and alkenylene groups having 1 to 4 double bonds each of which may be linear or branched and has 2 to 34 carbon atoms, particularly, alkylene groups and alkenylene groups having 1 to 4 double bonds each of which may be linear or branched and has 2 to 24 carbon atoms being preferred. Specific examples include ethylene, trimethylene, tetramethylene, hexamethylene, heptamethylene, octamethylene, decamethylene, undecamethylene, dodecamethylene, tridecamethylene, tetradecamethylene, hexadecamethylene, octadecamethylene, tricosamethylene, hexacosamethylene, triacontamethylene, 1-methylethylene, 2-ethyltrimethylene, 1-methylheptamethylene, 2-methylheptamethylene, 1-butylhexamethylene, 2-methyl-5-ethylheptamethylene, 2,3,6-trimethylheptamethylene, 6-ethyldecamethylene, 7-methyltetradecamethylene, 7-ethylhexadecamethylene, 7,12-dimethyloctadecamethylene, 8,11-dimethyloctadecamethylene, 7,10-dimethyl-7-ethylhexadecamethylene, 1-octadecylethylene, 9,10-dioctyloctadecamethylene, 8,9-dinonylhexadecamethylene, ethenylene, 1-octadecenylethylene, 7,11-octadecadienylene, 7-ethenyl-9-hexadecamethylene, 7,12-dimethyl-7,11-octadecadienylene, 8,11-dimethyl-7,11-octadecadienylene, 9,10-dioctyl-7,11-octadecadienylene and 8,9-dinonyl-6,10-hexadecadienylene groups. Of these, 7,12-dimethyloctadecamethylene, 7,12-dimethyl-7,11-octadecadienylene, octadecamethylene, octamethylene, decamethylene, undecamethylene and tridecamethylene groups are more preferred.

[0011] In the formula (2), preferred as R^{3a} are alkylene groups and alkenylene groups having 1 to 4 double bonds each of which may be linear or branched and has 12 to 34 carbon atoms, with alkylene groups and alkenylene groups having 1 to 4 double bonds each of which may be linear or branched and has 12 to 24 carbon atoms being more preferred. Of these, 7,12-dimethyloctadecamethylene, 7,12-dimethyl-7,11-octadecadienylene, octadecamethylene and tridecamethylene groups are still more preferred.

[0012] In the diamide derivatives (2) of the present invention, particularly preferred are compounds of the formula (2) having, as R^1 , R^2 and R^{3a} , the groups within the above-described more preferred ranges in combination, respectively. In the diamide derivatives (1) to be used for the dermatologic preparations of the present invention, particularly preferred are compounds of the formula (1) having, as R^1 , R^2 and R^3 , groups within the above-described more preferred ranges in combination, respectively.

[0013] Particularly preferred examples of the diamide derivatives (1) to be used for the dermatologic preparations of the present invention include:





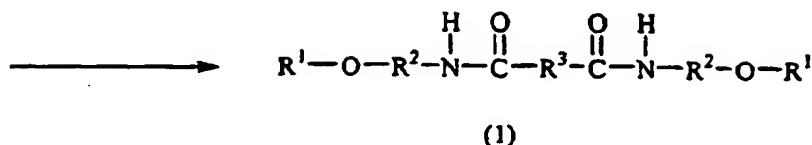
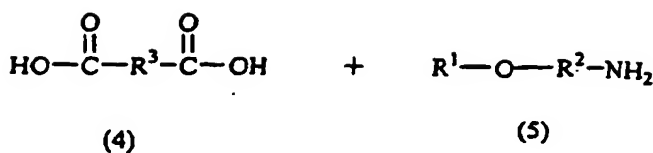
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[0014] Particularly preferred examples of the diamide derivatives (2) of the present invention include the above-exemplified compounds (A) to (C), (F) to (J) and (L).

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[0015] The diamide derivatives (1) to be used for the dermatologic preparations of the present invention can be prepared by a known amide synthesis method. A preparation process, for example, in accordance with the following reaction scheme can efficiently yield them at a low cost.

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(wherein, R¹, R² and R³ have the same meanings as described above).

[0016] The target diamide derivative (1) is available efficiently by condensing a corresponding carboxylic acid (4) or reactive derivative thereof (ester, acid halide or acid anhydride) with an amine (5). This condensation is preferably conducted in the presence or absence of a dehydrating agent such as dicyclohexylcarbodiimide, or a base, for example, an alkali metal hydroxide such as potassium hydroxide or sodium hydroxide, an alkaline earth metal hydroxide such as calcium hydroxide, an alkali metal carbonate such as potassium carbonate, an alkali earth metal carbonate such as calcium carbonate, an alkali metal alcoholate such as sodium methoxide, sodium ethoxide or potassium-tert-butoxide or a tertiary amine such triethylamine or pyridine under a pressure ranging from normal pressure to reduced pressure at 1.3 Pa at room temperature to 250°C. Upon condensation, the amine (5) is preferably used in an excess amount, more specifically, 2 equivalents or greater of the dicarboxylic acid (4) or reactive derivative thereof. For rapid progress of the reaction, it is preferred to conduct the reaction while removing, out of the system, water or an alcohol generated by the reaction. The diamide derivative (1) thus available can be purified in a known manner such as washing with water, column chromatography, distillation, crystallization, recrystallization or powder treatment. Since the diamide derivative (1) obtained in such a manner has effects of penetrating into the lipid layer of the horny layer, thereby maintaining and improving the water retention capacity and barrier functions of the horny layer, it is useful as a humectant or skin-barrier-function reinforcing agent.

[0017] The dermatologic preparations of the present invention are obtained by incorporating the amide derivative (1) in a base (carrier) ordinarily employed for dermatologic preparations. They can be prepared by mixing necessary raw materials in a known manner.

[0018] The dermatologic preparations of the present invention may be broadly classified into medicinal dermatologic preparations and cosmetics depending on the applications thereof. Examples of the medicinal dermatologic preparations include various ointments containing pharmaceutically effective ingredients. These ointments may contain either an oily base or an O/W or W/O emulsion base. No particular limitation is imposed on the oily base and examples thereof include vegetable oils, animal oils, synthetic oils, fatty acids and natural and synthetic glycerides. No particular limitation is imposed on the pharmaceutically effective ingredients and examples thereof include analgesic antiinflammatory agents, antipruritic agents, bactericides, astringents, skin emollients and hormones as needed.

[0019] When the dermatologic preparations of the present invention are used as cosmetics (including skin cosmetics and hair cosmetics), the essential ingredient, that is, the diamide derivative (1), may be arbitrarily blended with commonly used oleaginous components, surfactants, humectants, ultraviolet absorbers, whitening agents, anti-wrinkle compositions, alcohols, chelating agents, pH regulators, antiseptics, thickeners, colorants and perfumes.

[0020] These cosmetics may be formulated into various forms such as W/O and O/W emulsion cosmetics, cream, cosmetic milky lotion, cosmetic lotion, oily cosmetic, lipstick, foundation, bath agent, skin cleanser, nail treatment and hair cosmetics. No particular limitation is imposed on the hair cosmetics and examples include hair tonic, hair dressing, hair rinse, hair treatment, hair conditioner, hair styling agent, shampoo, hair nourishment and hair growth stimulant.

[0021] Although there is no particular limitation imposed on the content of the diamide derivative (1) in the dermatologic preparation of the present invention. In the case of an emulsion type dermatologic preparation, 0.001 to 50 wt. % (which will hereinafter be described % simply) based on the whole composition is preferred. In the case of an oily dermatologic preparation containing a liquid hydrocarbon such as squalane as a base, 0.01 to 50% based on the whole composition is preferred. In either case, 0.01 to 20% is particularly preferred. Particularly, for prevention or remedy of chapping, addition of 0.1 to 20% is preferred. Use of the dermatologic preparation as skin cosmetics is particularly preferred.

[0022] When the dermatologic preparations of the present invention serve as a medicinal dermatologic preparation or skin cosmetic, surfactants such as nonionic surfactants, anionic surfactants, cationic surfactants and amphoteric

surfactants may be incorporated. Of these, nonionic surfactants such as polyoxyethylene alkyl ethers, polyoxyethylene fatty acid esters, sorbitan fatty acid esters, polyoxyethylene sorbitan fatty acid esters, fatty acid monoglycerides and glyceryl ethers are preferred. As its content, 0.01 to 20%, particularly 0.1 to 10% in the whole preparation is preferred.

[0023] Although no particular limitation is imposed on the content of the diamide derivative (1) in the hair cosmetic of the present invention, preferred is 0.001 to 5% for shampoo, 0.1 to 20% for rinse, treatment, conditioner or styling agent and 0.01 to 5% for hair liquid or hair tonic.

[0024] The hair cosmetic of the present invention may contain surfactants such as anionic surfactants, cationic surfactants, nonionic surfactants and amphoteric surfactants and in addition, components ordinarily employed for hair cosmetics. When the hair cosmetic of the present invention is a shampoo, it may contain, as a main active agent, an anionic surfactant such as an alkyl ether sulfate, alkyl sulfate or olefin sulfonate. As its content, 5 to 30%, particularly 10 to 20%, each based on the whole composition, is preferred.

[0025] When the cosmetic of the present invention is a hair rinse, conditioner, hair treatment or hair styling agent, it may contain a cationic surfactant such as mono- or di-(long chain alkyl) tetraammonium salt, a nonionic surfactant such as polyoxyethylene alkyl or alkenyl ether or an oil/fat such as liquid paraffin for imparting the hair with good touch feel. The content of the cationic or nonionic surfactant is preferably 0.1 to 50%, particularly 0.5 to 20% in the whole composition.

[0026] When the hair cosmetic is a hair liquid or hair tonic, it may contain a nonionic surfactant such as polyoxyethylene. The nonionic surfactant is preferably added in an amount of 0.01 to 20%, particularly 0.1 to 5% in the whole composition.

[0027] The skin cosmetic or hair cosmetic of the present invention containing the diamide derivative (1) may be formulated into an aqueous solution, ethanol solution, emulsion, suspension, gel, solid, aerosol or powder without any limitation.

Examples

Preparation Example 1

Preparation of Compound (A)

[0028] In a flask equipped with a stirrer, a nitrogen inlet tube and a distilling tube, 150 g of dimethyl 8,13-dimethyleicosanedioate ("IPS-22M", product of Okamura Seiyu), 159 g of diglycolamine and 7.5 g of sodium methoxide were charged. The mixture was stirred for 5 hours at 140°C under reduced pressure (20 torr), while distilling off the methanol by-produced. After completion of the reaction, excess diglycolamine was distilled off under reduced pressure. The residue was then washed with water, whereby 200 g (yield: 98%) of the title compound was obtained. The resulting compound 1 (A) has following physical properties:

Colorless paste

¹H-NMR (CDCl₃, δ): 0.67-0.91 (m,6H), 0.93-1.58(m,26H), 1.59-1.76(m,4H), 2.17(t,J=7.2Hz,4H), 2.72-3.12(m,2H), 3.33-3.52(m,4H), 3.52-3.64(m,8H), 3.65-3.85(m,4H), 6.13-6.56(m,2H).

Preparation Examples 2 to 15

[0029] In a similar manner to Example 1 except for the use of the dicarboxylic acid (4) or reactive derivative thereof and, as the amine (5), the compound shown in Table 1 or 2, Compounds (B) to (O) were obtained. These raw materials are shown in Tables 1 and 2, together with the physical properties of the diamide derivatives.

Table 1

	Diamide derivative (1) prepared	Dicarboxylic acid (4) or reactive derivative thereof	Amine (5)	Physical properties
Prep. Ex. 2	Compound (B)	Dimethyl 7,12-dimethyl- 7,11-octadecadiene- 1,18-dicarboxylate ("IPU-22MM", product of Okamura Seiyu)	Diglycolamine	Colorless paste ¹ H-NMR (CDCl ₃ , δ): 1.05-1.45 (m,16H), 1.45-1.72 (m,8H), 1.79-2.08 (m,6H), 2.14 (t,J=7.8Hz,4H), 2.83-3.15 (m,2H), 3.32-3.47 (m,4H), 3.47-3.61 (m,8H), 3.61-3.78 (m,4H), 4.52-5.19 (m,2H), 6.26-6.58 (m,2H).
Prep. Ex. 3	Compound (C)	Dimethyl eicosanedioate ("SL- 20MM", product of Okamura Seiyu)	Diglycolamine	White crystals, Melting point: 135°C ¹ H-NMR (MeOH-d ₄ , δ): 1.18-1.44 (m,28H), 1.48-1.71 (m,4H), 2.18 (t,J=7.4Hz,4H), 3.31-3.43 (m,4H), 3.46-3.60(m,8H), 3.60-3.71(m,4H), 4.49-4.68(m,2H).
Prep. Ex. 4	Compound (D)	Dimethyl sebacate	Diglycolamine	White solid, Melting point: 107°C ¹ H-NMR (CDCl ₃ , δ): 1.22-1.40(m,8H), 1.54-1.72(m,4H), 2.19 (t,J=7.2Hz,4H), 2.70-2.90 (m,2H), 3.36-3.54 (m,4H), 3.54-3.70 (m,8H), 3.70-3.89 (m,4H), 6.13-6.30 (m,2H).
Prep. Ex. 5	Compound (E)	Dimethyl succinate	Diglycolamine	White crystals, Melting point: 85°C ¹ H-NMR (CDCl ₃ , δ): 2.52 (s,4H), 3.30-3.50 (m,4H), 3.50-3.64 (m, 8H), 3.64-3.88 (m, 4H), 7.12-7.33 (m, 2H).

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Table 1 (continued)

	Diamide derivative (1) prepared	Dicarboxylic acid (4) or reactive derivative thereof	Amine (5)	Physical properties
Prep. Ex. 6	Compound (F)	Dimethyl 8,13-dimethyleicosane- dioate ("IPS-22MM", product of Okamura Seiyu)	3-Methoxy- propylamine	Colorless paste ¹ H-NMR (CDCl ₃ , δ): 0.67-0.95(m,6H), 1.00-1.48 (m,26H), 1.48-1.70 (m,4H), 1.70-1.88 (m,4H), 2.14 (t,J=7.2Hz,4H), 3.23-3.41 (m,10H), 3.41-3.58 (m,4H), 5.96-6.20 (m,2H).
Prep. Ex. 7	Compound (G)	Dimethyl 8,13-dimethyleicosane- dioate ("IPS-22MM", product of Okamura Seiyu)	3-Dodecyloxy- propylamine	White crystals, Melting point: 36°C ¹ H-NMR (CDCl ₃ , δ): 0.64-0.98 (m,12H), 0.98-1.46 (m,62H), 1.46-1.70 (m,8H), 1.70-1.88 (m,4H), 2.14 (t,J=7.2Hz,4H), 3.26-3.47 (m,8H), 3.47-3.61 (m,4H), 6.09-6.34 (m,2H).
Prep. Ex. 8	Compound (H)	Dimethyl 8,13-dimethyleicosane- dioate ("IPS-22MM", product of Okamura Seiyu)	2-(2-Methoxy- ethoxy)-ethylamine	Colorless paste ¹ H-NMR (CDCl ₃ , δ): 0.69-0.88(m,6H), 0.93-1.41 (m,26H), 1.49-1.68(m,4H), 2.27 (t,J=7.2Hz,4H), 3.36 (s,6H), 3.40-3.68(m, 16H), 5.88-6.06(m, 2H).

Table 2

	Diamide derivative (1) prepared	Dicarboxylic acid (4) or reactive derivative thereof	Amine (5)	Physical properties
Prep. Ex. 9	Compound (I)	Dimethyl pentadecanedioate	Diglycolamine	White crystals, Melting point: 127°C ¹ H-NMR (MeOH-d ₄ , δ): 1.20-1.45 (m,18H), 1.45-1.75 (m,4H), 2.18 (t,J=7.4Hz,4H), 3.25-3.45 (m,8H), 3.52 (t,J=5H.1Hz,4H), 3.60-3.75 (m,4H), 4.60-4.70(m,2H).

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Table 2 (continued)

	Diamide derivative (1) prepared	Dicarboxylic acid (4) or reactive derivative thereof	Amine (5)	Physical properties
Prep. Ex.10	Compound (J)	Dimethyl brassylate	Diglycolamine	White crystals, Melting point: 120°C ¹ H-NMR (MeOH-d ₄ , δ): 1.20-1.45 (m,14H), 1.50-1.70 (m,4H), 2.18 (t,J=7.4Hz,4H), 3.25-3.45 (m,8H), 3.52 (t,J=5.0Hz,4H), 3.60-3.75 (m,4H), 4.60-4.70 (m,2H).
Prep. Ex.11	Compound (K)	Dimethyl dodecanedioate	Diglycolamine	White solid, Melting point: 118°C ¹ H-NMR (MeOH-d ₄ , δ): 1.20-1.45 (m,12H), 1.45-1.75 (m,4H), 2.18 (t,J=7.4Hz,4H), 3.25-3.45 (m,8H), 3.52 (t,J=5.1Hz,4H), 3.60-3.75 (m,4H), 4.60-4.70 (m,2H).
Prep. Ex.12	Compound (L)	Dimethyl Eicosanedioate	3-Methoxy- propylamine	White crystals, Melting point: 134 °C ¹ H-NMR (CDCl ₃ , δ): 1.15-1.40 (m,32H), 1.52-1.70 (m,4H), 1.77 (quintet, J=6.1Hz,4H), 2.15 (t,J=7.6Hz,4H), 3.30-3.43 (m,4H), 3.35 (s,6H), 3.48 (t, J=5.7Hz,4H), 5.90-6.10 (m,2H).
Prep. Ex. 13	Compound (M)	Dimethyl dodecanedioate	3-Methoxy- propylamine	White crystals, Melting point: 129°C ¹ H-NMR (CDCl ₃ , δ): 1.15-1.40 (m, 12H), 1.50-1.70(m,4H), 1.70-1.90 (m,4H), 2.15 (t,J=6.1Hz,4H), 3.30-3.45 (m,4H), 3.35 (s,6H), 3.48 (t, J=5.8Hz,4H), 5.95-6.05 (m,2H).

Table 2 (continued)

	Diamide derivative (1) prepared	Dicarboxylic acid (4) or reactive derivative thereof	Amine (5)	Physical properties
Prep. Ex. 14	Compound (N)	Dimethyl sebacate	3-Methoxy- propylamine	White crystals, Melting point: 125°C ¹ H-NMR (CDCl ₃ δ): 1.20-1.45 (m,8H), 1.50-1.70 (m,4H), 1.71-1.90 (m,4H), 2.15 (t,J=7.5Hz,4H), 3.25-3.45 (m,4H), 3.35 (s,6H), 3.48 (t, J=5.8Hz,4H), 6.00-6.20 (m,2H).
Prep. Ex.15	Compound (O)	Dimethyl sebacate	3-Dodecyloxy- propylamine	White crystals, Melting point: 134°C ¹ H-NMR (CDCl ₃ , δ): 0.80-1.00 (m,6H), 1.15-1.45 (m,44H), 1.45-1.70 (m,8H), 1.76 (quintet, J=6.1Hz,4H), 2.13 (t,J=7.6Hz,4H), 3.30-3.48 (m,8H), 3.52 (t,J=5.7Hz,4H), 6.05-6.25 (m,2H)

Preparation Example 16

[0030] In a flask equipped with a stirrer, a nitrogen inlet tube and a distilling tube, 537 g of 8,13-dimethyl-eicosanedioic acid ("IPS-22", product of Okamura Seiyu) was charged, followed by dropwise addition of 381 g of diglycolamine over 4 hours under stirring at 180°C in a nitrogen gas stream. After saturation for 4 hours under the same conditions, saturation was conducted for further 4 hours at 200°C. The above-described dropwise addition and saturation were conducted in a nitrogen gas stream while distilling off water by-produced. After completion of the reaction, excess diglycolamine was removed by distillation under reduced pressure and steaming. In a molecular distillation apparatus, a low-boiling-point substance was removed under the conditions of 200°C and 0.7 Pa, whereby 680 g (yield: 86%) of the crude Compound (A) was obtained as a pale yellow paste. Mainly for bleaching, 20 g of the compound was dissolved in 20.0 g of ethanol and 0.40 g of activated charcoal was added to the resulting solution, followed by stirring at 80°C for 2 hours. The activated charcoal was filtered out and the residue was concentrated under reduced pressure, whereby 19.9 g of Compound (A) was obtained as a colorless paste.

Preparation Example 17

[0031] In a flask equipped with a stirrer, a nitrogen inlet tube and a distilling tube, 770 g of sebacic acid was charged, followed by dropwise addition of 1001 g of diglycolamine over 4 hours under stirring at 200°C in a nitrogen gas stream. The reaction mixture was saturated for 5 hours under the same conditions. The above-described dropwise addition and saturation were conducted in a nitrogen gas stream while distilling off water by-produced. After completion of the reaction, excess diglycolamine was removed by distillation under reduced pressure and steaming, whereby 1410 g (yield: 98%) of the crude Compound (D) was obtained as white crystals. The crystals were recrystallized from 4510 g of a 6.25% aqueous solution of sodium sulfate, whereby 1085 g (yield: 76%) of Compound (D) was obtained as white crystals.

Preparation Example 18

[0032] In a flask equipped with a stirrer, a nitrogen inlet tube and a distilling tube, 100 g of 8,13-dimethyl-eicosanedioic

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acid ("IPS-22", product of Okamura Seiyu) was charged, followed by the dropwise addition of 61 g of 3-methoxypropylamine over 3 hours at 180°C under stirring in a nitrogen gas stream. After saturation for 3 hours under the same conditions, saturation was conducted for further 5 hours at 200°C. The above-described dropwise addition and saturation were conducted under a nitrogen gas stream while distilling off water by-produced. After completion of the reaction, excess 3-methoxypropylamine was removed by distillation under reduced pressure and steaming, whereby 137 g (yield: 99%) of the crude Compound (F) was obtained as a pale yellow paste. The crude Compound (F) was distilled (at 220°C and 0.7 to 0.3 Pa), whereby 106 g (yield: 77%) of Compound (F) was obtained as a colorless paste.

Test 1

[0033] Dermatologic preparations (invention products) composed of 33% of each of the diamide derivatives prepared in Preparation Examples 1 to 18 and 67% of vaseline were prepared. Their effects on skin conductance and chapping were evaluated in the below-described manner. For comparison, a dermatologic preparation (Comparative Product 1) composed of vaseline alone was evaluated in the same manner. Table 3 shows the results.

(Test method)

[0034] Ten female subjects aged 20 to 50 years, who suffered chapping in cheeks in winter, were employed. Different dermatologic preparations were applied on the right and left cheeks of the subjects for 2 weeks, respectively. On the next day after completion of the application for 2 weeks, the following items were examined.

(1) Skin conductance

The face of each subject was washed with warm water of 37°C. After she was allowed to stand in a room at 20°C under a humidity of 45% for 20 minutes, the water content of the horny layer was measured using a skin conductance meter (product of IBS Co., Ltd.).

(2) Skin chapping score

Skin chapping was observed with naked eyes and evaluated based on the following criteria. Each score is expressed in the mean value.

- 0: No chapping is observed.
- 1: Slight chapping is observed.
- 2: Chapping is observed.
- 3: Somewhat serious chapping is observed.
- 4: Serious chapping is observed.

Table 3

Dermatologic preparation	Diamide derivative	Skin conductance	Skin chapping score
Invention product 1	Compound (A)	28	0.6
Invention product 2	Compound (B)	25	0.9
Invention product 3	Compound (C)	21	1.0
Invention product 4	Compound (D)	30	0.5
Invention product 5	Compound (E)	18	1.0
Invention product 6	Compound (F)	20	1.2
Invention product 7	Compound (G)	17	1.1
Invention product 8	Compound (H)	19	1.3
Invention product 9	Compound (J)	23	0.8
Invention product 10	Compound (L)	19	1.3
Invention product 11	Compound (N)	18	1.5
Comparative product 1 (vaseline alone)	-	5	2.7

[0035] The results show that the invention products 1 to 11 were superior to the comparative product in effects of increasing the water content of the horny layer and remedying skin chapping.

Test 2

[0036] Dermatologic preparations (invention products) of the present invention composed of 10% of each of the diamide derivatives prepared in Preparation Examples 1 to 18 and 90% of squalane were prepared and their transepidermal water loss and percutaneous absorption were evaluated by the below-described test method. For comparison, a dermatologic preparation (Comparative Product 2) composed of squalane alone was also evaluated by the test. Table 4 shows the results.

(Test method)

[0037] Wistar male rats were fed with a feed free from essential fatty acids and the resulting rats with essential fatty acid deficiency were employed in this test. The dorsal part of the rat showing essential fatty acid deficiency was carefully shaven and each dermatologic preparation to be evaluated was applied thereto once a day for 2 weeks. A group consisting of 5 rats was provided for the test of each dermatologic preparation. After 2 weeks, the following items were examined.

(1) Transepidermal water loss

The dorsal part of the test rat was washed with warm water and the animal was allowed to stand in a room (at 23°C under a humidity of 45%) for 1 hour. The transepidermal water loss was then measured with an evaporimeter. Each measured value is expressed in mean value.

(2) Percutaneous water absorption

After the dorsal skin of the rat was washed with warm water of 37°C, it was cut and fixed in a percutaneous absorption chamber with the epidermal side thereof directed upward. A lower receiver of the chamber was filled with a phosphate buffer solution while a container on the epidermal side thereof was charged with 1 ml of a phosphate buffer solution containing 37KBq of ¹⁴C-salicylic acid. Two hours after they were allowed to stand, 1 mL of the phosphate buffer solution was taken out of the lower receiver and the radioactivity of the ¹⁴C-salicylic acid which had penetrated into it was measured. Each value is expressed in mean value.

Table 4

Dermatologic preparation	Diamide derivative	Transepidermal water loss	Percutaneous absorption
Invention product 12	Compound (A)	12	572
Invention product 13	Compound (B)	12	558
Invention product 14	Compound (C)	21	1685
Invention product 15	Compound (D)	15	997
Invention product 16	Compound (E)	25	1772
Invention product 17	Compound (F)	20	1487
Invention product 18	Compound (G)	16	1096
Invention product 19	Compound (H)	23	2036
Invention product 20	Compound (I)	18	1197
Invention product 21	Compound (J)	21	1490
Invention product 22	Compound (K)	22	1694
Invention Product 23	Compound (L)	15	705
Invention Product 24	Compound (M)	25	2038
Invention Product 25	Compound (N)	23	1560
Invention Product 26	Compound (O)	27	2476

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Table 4 (continued)

Dermatologic preparation	Diamide derivative	Transepidermal water loss	Percutaneous absorption
Comparative product 2 (squalane alone)	-	31	2994

[0038] The results show that the invention products 12 to 26 were superior to the comparative product 2 in effects of suppressing transepidermal water loss and percutaneous absorption and remedying chapping.

Test 3

[0039] Hair rinses having the compositions as listed in Table 5 were prepared and excessive dryness and touch feel of the hair after treated with the resulting hair rinses were evaluated by a panel of 5 experts in accordance with the below-described criteria. The results are shown in Table 6.

(Evaluation criteria)

[0040]

- 2: Bad
- 1: slightly bad
- 0: neither good nor bad
- +1: slightly good
- +2: Good

Table 5

Composition (%)	Invention product 27	Invention Product 28	Invention Product 29	Comparative Product 3
Distearyl dimethyl ammonium chloride	2	2	2	2
Propylene glycol	3	3	3	3
Compound (A)	1	-	-	-
Compound (D)	-	1	-	-
Compound (F)	-	-	1	-
Water	Balance	Balance	Balance	Balance

Table 6

Evaluation items	Invention product 27	Invention product 28	Invention product 29	Comparative Product 3
Excessive dryness of the hair	+1.4	+1.2	+1.6	-0.4
Favorable touch feel	+1.2	+1.6	+1.8	-0.5

[0041] Apparent from the above-described table, the invention products 27 to 29 are superior to the comparative product 3 in the effects of alleviating excessive hair dryness and improving the touch feel of the hair.

Example 1

[0042] A skin lotion having the composition as shown in Table 7 was prepared in a conventional manner. The resulting hair lotion exhibited excellent effects of preventing or remedying chapping. Compound (A) and Compound (F) were excellent in miscibility and mixing stability.

Table 7

(Composition)	(%)
Compound (A) or Compound (F)	1
Glycerin monostearate	1
Ethanol	15
Propylene glycol	4
Isopropyl palmitate	3
Lanolin	1
Methyl paraoxybenzoate	0.1
Ceramide	1
Perfume	Trace
Colorant	Trace
Water	Balance

Example 2

[0043] An O/W cream having the composition as shown in Table 8 was prepared in a conventional manner. The resulting O/W cream exhibited excellent effects of preventing or remedying chapping. Compound (D) and Compound (F) were excellent in miscibility and mixing stability.

Table 8

(Composition)	(%)
Compound (D) or Compound (F)	3.5
Squalane	2.0
Neopentyl glycol dicaprate	3.0
Polyoxyethylene (20) sorbitan monostearate	2.1
Sorbitan monostearate	0.9
Polyoxyethylene hardened castor oil ("EMANON CH-40", product of Kao Corp.)	1.0
Monoisostearyl glyceryl ether	0.2
86% glycerin	5.0
Methyl paraben	0.3
Water	Balance

Example 3

[0044] A shampoo having the composition as shown in Table 9 was prepared in a conventional manner. This shampoo improved the touch feel of the hair and prevented remedied the chapping of the scalp. Compound (F) and Compound (A) were excellent in miscibility and mixing stability.

Table 9

(Composition)	(%)
Polyoxyethylene (25) lauryl ether sulfate Sodium salt	15
Coconut oil fatty acid diethanol amide	3
Compound (A) or Compound (F)	2
Perfume	0.5
Colorant	Trace
Citric acid	Trace
Water	Balance

Example 4

[0045] A hair liquid composition having the composition as shown in Table 10 was prepared in a conventional manner. The resulting hair liquid composition imparted the hair with excellent style retention and manageability and good touch feel. Compound (D) and Compound (F) were excellent in miscibility and mixing stability.

Table 10

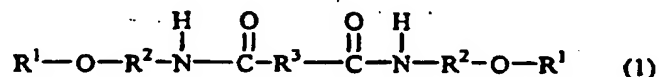
(Composition)	(%)
Compound (D) or Compound (F)	1
Polyoxypropylene (30) butyl ether	15.0
Ethanol	40.0
Water	Balance
Perfume	0.3

Industrial Applicability

[0046] Diamide derivatives (I) penetrate the lipid layer between horny cells, thereby exerting effects of improving (maintaining · reinforcing) water retention capacity and barrier functions of the horny layer; and by these effects, they prevent or remedy the skin chapping and prevent aging of the skin. When they are applied to the hair, they penetrate into the hair, thereby heightening their protective effects, improving the touch feel of the hair and preventing or remedying the chapping of the scalp. In addition, they have good solubility and stability in a base and have excellent mixing stability, making it possible to prepare a dermatological preparation efficiently at a low cost.

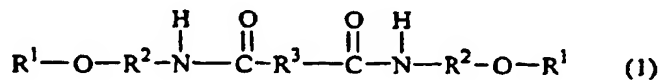
Claims

1. A dermatologic preparation, which comprises a diamide derivative represented by the following formula (1):



(wherein, R¹ represents a linear or branched hydrocarbon group having 1 to 22 carbon atoms which may be substituted by one or more hydroxy and/or alkoxy groups, R² represents a linear or branched divalent hydrocarbon group having 1 to 12 carbon atoms, and R³ represents a linear or branched divalent hydrocarbon group having 1 to 42 carbon atoms).

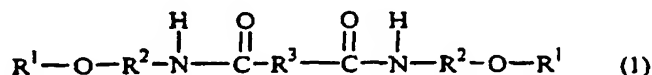
2. A dermatologic preparation according to claim 1, which is a cosmetic preparation.
3. A dermatologic preparation as claimed in claim 1 or 2, wherein R¹ represents a linear or branched alkyl group having 1 to 22 carbon atoms which may have 1 to 3 substituents selected from a hydroxy group and C₁₋₆ alkoxy groups, R² represents a linear or branched hydrocarbon group having 1 to 12 carbon atoms, and R³ represents an alkylene group or an alkenylene group having 1 to 4 double bonds, which alkylene or alkenylene group may be linear or branched and has 2 to 34 carbon atoms.
4. A humectant comprising, as an effective ingredient, a diamide derivative represented by the following formula (1):



(wherein, R¹ represents a linear or branched hydrocarbon group having 1 to 22 carbon atoms which may be substituted by one or more hydroxy and/or alkoxy groups, R² represents a linear or branched divalent hydrocarbon

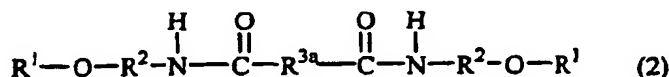
group having 1 to 12 carbon atoms, and R³ represents a linear or branched divalent hydrocarbon group having 1 to 42 carbon atoms).

5. A skin barrier function reinforcing agent comprising, as an effective ingredient, a diamide derivative represented by the following formula (1) :



(wherein, R¹ represents a linear or branched hydrocarbon group having 1 to 22 carbon atoms which may be substituted by one or more hydroxy and/or alkoxy groups, R² represents a linear or branched divalent hydrocarbon group having 1 to 12 carbon atoms, and R³ represents a linear or branched divalent hydrocarbon group having 1 to 42 carbon atoms).

6. A diamide derivative represented by the following formula (2):



(wherein, R¹ represents a linear or branched hydrocarbon group having 1 to 22 carbon atoms which may be substituted by one or more hydroxy and/or alkoxy groups, R² represents a linear or branched divalent hydrocarbon group having 1 to 12 carbon atoms, and R^{3a} represents an alkylene group or an alkenylene group having 1 to 4 double bonds, which alkylene or alkenylene group may be linear or branched and has 11 to 42 carbon atoms).

7. A diamide derivative according to claim 6, wherein R¹ represents a linear or branched alkyl group having 1 to 22 carbon atoms which may have 1 to 3 substituents selected from a hydroxy group and C₁₋₆ alkoxy groups, R² represents a linear or branched alkylene group having 1 to 12 carbon atoms, and R^{3a} represents an alkylene group or an alkenylene group having 1 to 4 double bonds, which alkylene or alkenylene group may be linear or branched and has 12 to 34 carbon atoms.

INTERNATIONAL SEARCH REPORT

International application No.

PCT/JP00/01383

A. CLASSIFICATION OF SUBJECT MATTER Int.Cl. ⁷ A61K 7/48, 7/00, 7/06, C07C 233/18		
According to International Patent Classification (IPC) or to both national classification and IPC		
B. FIELDS SEARCHED Minimum documentation searched (classification system followed by classification symbols) Int.Cl. ⁷ A61K 7/00-7/50, C07C 233/18		
Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched		
Electronic data base consulted during the international search (name of data base and, where practicable, search terms used) CA(STN), REGISTRY(STN), WPIDS(STN)		
C. DOCUMENTS CONSIDERED TO BE RELEVANT		
Category*	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
X	US, 3773056, A (Kalopissis et al.), 20 November, 1973 (20.11.73), Columns 3, 9, 11, lines 15 to 26 (Family: none)	1-3
A		4-7
A	JP, 10-218849, A (Kao Corporation), 18 August, 1998 (18.08.98), Claims (Family: none)	1-7
<input type="checkbox"/> Further documents are listed in the continuation of Box C. <input type="checkbox"/> See patent family annex.		
* Special categories of cited documents: "A" document defining the general state of the art which is not considered to be of particular relevance "E" earlier document but published on or after the international filing date "L" document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified) "O" document referring to an oral disclosure, use, exhibition or other means "P" document published prior to the international filing date but later than the priority date claimed "T" later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention "X" document of particular relevance; the claimed invention cannot be considered novel or cannot be considered to involve an inventive step when the document is taken alone "Y" document of particular relevance; the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art "&" document member of the same patent family		
Date of the actual completion of the international search 10 May, 2000 (10.05.00)		Date of mailing of the international search report 23 May, 2000 (23.05.00)
Name and mailing address of the ISA/ Japanese Patent Office		Authorized officer
Facsimile No.		Telephone No.

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